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FILE COVERS 1907 - 28 Feb 2002 VOL 136 ISS 9
FILE LAST UPDATED: 26 Feb 2002 (20020226/ED)

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```
=> s thiol?(w)chitosan
 67200 THIOL?
 12013 CHITOSAN
  629 CHITOSANS
 12031 CHITOSAN
      (CHITOSAN OR CHITOSANS)
L1      3 THIOL?(W)CHITOSAN
```

=> d L1 1-3 ti

L1 ANSWER 1 OF 3 CAPLUS COPYRIGHT 2002 ACS
TI Synthesis and in vitro evaluation of chitosan-thioglycolic acid conjugates

L1 ANSWER 2 OF 3 CAPLUS COPYRIGHT 2002 ACS
TI Thiolated polymers - thiomers: development and in vitro evaluation of chitosan-thioglycolic acid conjugates

L1 ANSWER 3 OF 3 CAPLUS COPYRIGHT 2002 ACS
TI Synthesis and in vitro evaluation of chitosan-cysteine conjugates

=> d L1 1-3 ibib,abs

L1 ANSWER 1 OF 3 CAPLUS COPYRIGHT 2002 ACS
ACCESSION NUMBER: 2001:600181 CAPLUS
TITLE: Synthesis and in vitro evaluation of chitosan-thioglycolic acid conjugates
AUTHOR(S): Bernkop-Schnurch, Andreas; Hopf, Thorid E.
CORPORATE SOURCE: Institute of Pharmaceutical Technology and Biopharmaceutics, Center of Pharmacy, University of Vienna, Vienna, A-1090, Austria
SOURCE: Sci. Pharm. (2001), 69(2), 109-118
CODEN: SCPHA4; ISSN: 0036-8709
PUBLISHER: Oesterreichische Apotheker-Verlagsgesellschaft
DOCUMENT TYPE: Journal
LANGUAGE: English

AB The cationic thiomer chitosan-thioglycolic acid (TGA) shows excellent mucoadhesive features. In order to deepen the knowledge concerning this new excipient the optimization of its synthesis and a detailed characterization of its properties was the objective of this study. Mediated by increasing quantities of a carbodiimide, thioglycolic acid was covalently attached to chitosan forming amide bonds with the primary amino

groups of the polymer Dated. with Ellman's reagent, 38 .+- . 3, 104 .+- . 2, 685 .+- . 43, and 885 .+- . 7 .mu.mol thiol groups (n=3, .+- . SD) were bound

per g polymer at carbodiimide concns. of 50, 75, 100, and 125 mM, resp. The immobilized thiol groups displayed a comparatively higher reactivity to form disulfide bonds than the thiol groups in a corresponding mixt. of chitosan and free unconjugated TGA. In an aq. 0.5% (m/v) chitosan-TGA gel

59 .+- . 5% of the thiol groups formed disulfide bonds within 6 h at pH 6.0, whereas merely 5 .+- . 3% were oxidized in the corresponding phys. mixt. of chitosan and TGA. Diffusion studies showed that the modified polymer was capable of binding cysteine and cysteine Me ester. The result

supports the theory that the improved mucoadhesive properties of thiolated chitosan are based on the formation of disulfide bonds with cysteine moieties of mucus glycoproteins. Because of

its availability via an efficient synthetic pathway and its mucoadhesive properties based on the capability to bind cysteine subunits, chitosan-TGA

seems to be a promising new excipient for various drug delivery systems.
REFERENCE COUNT: 17 THERE ARE 17 CITED REFERENCES AVAILABLE FOR THIS

FORMAT RECORD. ALL CITATIONS AVAILABLE IN THE RE

L1 ANSWER 2 OF 3 CAPLUS COPYRIGHT 2002 ACS
ACCESSION NUMBER: 2001:520176 CAPLUS
TITLE: Thiolated polymers - thiomers: development and in vitro evaluation of chitosan-thioglycolic acid conjugates
AUTHOR(S): Kast, C. E.; Bernkop-Schnurch, A.
CORPORATE SOURCE: Center of Pharmacy, Institute of Pharmaceutical

Technology and Biopharmaceutics, University of
Vienna,

SOURCE: Vienna, A-1090, Austria
Biomaterials (2001), 22(17), 2345-2352

PUBLISHER: CODEN: BIMADU; ISSN: 0142-9612
Elsevier Science Ltd.

DOCUMENT TYPE: Journal
LANGUAGE: English

AB The aim of this study was to improve mucoadhesive properties of chitosan by the covalent attachment of thiol moieties to this cationic polymer. Mediated by a carbodiimide, thioglycolic acid (TGA) was covalently attached to chitosan. This was achieved by the formation of amide bonds between the primary amino groups of the polymer and the carboxylic acid group of TGA. Dependent on the pH-value and the wt. ratio of polymer to TGA during the coupling reaction the resulting thiolated polymers, the so-called thiomers, displayed 6.58, 9.88, 27.44, and 38.23 .mu.mole thiol groups per g polymer. Tensile studies carried out with these chitosan-TGA

conjugates on freshly excised porcine intestinal mucosa demonstrated a 6.3-, 8.6-, 8.9-, and 10.3-fold increase in the total work of adhesion (TWA) compared to the unmodified polymer, resp. In contrast, the combination of chitosan and free unconjugated TGA showed almost no mucoadhesion. These data were in good correlation with further results obtained by another mucoadhesion test demonstrating a prolonged residence time of **thiolated chitosan** on porcine mucosa. The swelling behavior of all conjugates was thereby exactly in the same range as for an unmodified polymer pretreated in the same way. Furthermore, it could be shown that chitosan-TGA conjugates are still biodegradable by

the glycosidase lysozyme. According to these results, chitosan-TGA conjugates represent a promising tool for the development of mucoadhesive drug delivery systems.

REFERENCE COUNT: 24 THERE ARE 24 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE

FORMAT

L1 ANSWER 3 OF 3 CAPLUS COPYRIGHT 2002 ACS
ACCESSION NUMBER: 2000:31626 CAPLUS
DOCUMENT NUMBER: 132:98016
TITLE: Synthesis and in vitro evaluation of
chitosan-cysteine conjugates

AUTHOR(S): Bernkop-Schnurch, Andreas; Brandt, Ursula-Maria;
Clausen, Andreas E.

CORPORATE SOURCE: Institut Pharmazeutische Technologie,
Pharmazie-Zentrum, Univ. Wien, Vienna, A-1090,

Austria

SOURCE: Sci. Pharm. (1999), 67(4), 197-208
CODEN: SCPHA4; ISSN: 0036-8709

PUBLISHER: Oesterreichische Apotheker-Verlagsgesellschaft
DOCUMENT TYPE: Journal
LANGUAGE: German

AB Mediated by a water-sol. carbodiimide cysteine was covalently attached to chitosan. According to the amt. of carbodiimide during the coupling reaction, 0.25, 0.7, and 1.2% of Cys were thereby bound to the polymer. Whereas the mucoadhesive properties of chitosan could not be improved due to this modification, the stability of matrix tablets based on **thiolated chitosan** might be strongly improved because of

the formation of inter- and/or intramol. disulfide bonds within these polymers. This oxidative process can be accelerated at higher temps. and by lowering the proton concn. on the polymer. Permeation studies carried out by chambers with freshly excised intestinal mucosa from guinea pigs demonstrated furthermore an improved permeation enhancing effect of chitosan due to the covalent attachment of Cys on it.

REFERENCE COUNT: 11 THERE ARE 11 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE

FORMAT

=> s thiol?(w)pectin
67200 THIOL?
17317 PECTIN
4168 PECTINS
18878 PECTIN
(PECTIN OR PECTINS)
L2 0 THIOL?(W)PECTIN

=> s thiol?(w)hyaluronic
67200 THIOL?
10110 HYALURONIC
1 HYALURONICS
10110 HYALURONIC
(HYALURONIC OR HYALURONICS)
L3 0 THIOL?(W)HYALURONIC

=> s thiol?(w)carboxymethylcellulose
67200 THIOL?
5091 CARBOXYMETHYLCELLULOSE
48 CARBOXYMETHYLCELLULOSES
5111 CARBOXYMETHYLCELLULOSE
(CARBOXYMETHYLCELLULOSE OR CARBOXYMETHYLCELLULOSES)
L4 1 THIOL?(W)CARBOXYMETHYLCELLULOSE

=>

=> d L4 ti

L4 ANSWER 1 OF 1 CAPLUS COPYRIGHT 2002 ACS
TI **Thiolated carboxymethylcellulose:** in vitro evaluation
of its permeation enhancing effect on peptide drugs

=> d 14 ibib

L4 ANSWER 1 OF 1 CAPLUS COPYRIGHT 2002 ACS
ACCESSION NUMBER: 2001:24250 CAPLUS
DOCUMENT NUMBER: 135:308715
TITLE: **Thiolated carboxymethylcellulose:**
effect in vitro evaluation of its permeation enhancing
on peptide drugs
AUTHOR(S): Clausen, A. E.; Bernkop-Schnurch, A.
CORPORATE SOURCE: Institute of Pharmaceutical Technology and
Biopharmaceutics, Centre of Pharmacy, University of
Vienna, Austria
SOURCE: Eur. J. Pharm. Biopharm. (2001), 51(1), 25-32
CODEN: EJPBEL; ISSN: 0939-6411

PUBLISHER: Elsevier Science Ireland Ltd.
DOCUMENT TYPE: Journal
LANGUAGE: English
REFERENCE COUNT: 28 THERE ARE 28 CITED REFERENCES AVAILABLE FOR
THIS
FORMAT RECORD. ALL CITATIONS AVAILABLE IN THE RE

=> s thiol? (w)hydroxypropylcellulose
67200 THIOL?
1581 HYDROXYPROPYLCELLULOSE
4 HYDROXYPROPYLCELLULOSES
1582 HYDROXYPROPYLCELLULOSE
(HYDROXYPROPYLCELLULOSE OR HYDROXYPROPYLCELLULOSES)
L5 0 THIOL? (W)HYDROXYPROPYLCELLULOSE

=> s thiol? and hydroxypropylcellulose
67200 THIOL?
1581 HYDROXYPROPYLCELLULOSE
4 HYDROXYPROPYLCELLULOSES
1582 HYDROXYPROPYLCELLULOSE
(HYDROXYPROPYLCELLULOSE OR HYDROXYPROPYLCELLULOSES)
L6 3 THIOL? AND HYDROXYPROPYLCELLULOSE

=> d L6 1-3 ti, kwic

L6 ANSWER 1 OF 3 CAPLUS COPYRIGHT 2002 ACS
TI Curly hair-straightening composition comprising keratin-reducing substances and alcohols
AB . . . sufficiently straighten natural curly or frizzy hair without damaging the hair. Thus, a 1-pack product contained NaHSO3 2.0, 2-methyl-2,4-pentanediol 30.0, **hydroxypropylcellulose** 1.5, water 64.0 wt.%, and monoethanolamine to adjust the pH to 9.5.
ST hair straightening compn **thiol** alc

L6 ANSWER 2 OF 3 CAPLUS COPYRIGHT 2002 ACS
TI Tetrahydro-5-oxo-2-(3-oxo-2-isoxazolidinyl)-2-furancarboxylates
AB . . . bacteria, e.g., *Staphylococcus aureus* with min. inhibitory concn. of 6.25 .mu.g/mL. Tablets contg. II 300, corn starch 50, lactose 28, **hydroxypropylcellulose** L 20 and Mg stearate 2 mg were prep'd.
IT 75-08-1, Ethanethiol 108-98-5P, Thiophenol, preparation
RL: RCT (Reactant)
(thiolation by, of bromooxoglutaric acid)
IT 89469-94-3
RL: RCT (Reactant)
(thiolation of, by thiophenol)

L6 ANSWER 3 OF 3 CAPLUS COPYRIGHT 2002 ACS
TI **Thiol** derivatives of cellulose as supports for the immobilization of non-**thiol** enzymes
TI **Thiol** derivatives of cellulose as supports for the immobilization of non-**thiol** enzymes
AB . . . were introduced into cellulose in 2 ways. Chlorodeoxycellulose gave with Na2S2O3 a thiosulfate deriv., which was reduced to mercaptocellulose. Similarly, 3-chloro-2-**hydroxypropylcellulose** was converted to epoxide, which on reaction with Na2S2O2 and redn. gave 3-mercato-2-**hydroxypropylcellulose**. Acetylcholinesterase (EC 3.1.1.7) from bovine erythrocytes and from elec. eel,

butyrylcholinesterase (EC 3.1.1.8), and trypsin (EC 3.4.21.4) were immobilized on. . .

ST enzyme immobilization cellulose **thiol** deriv;
acetylcholinesterase immobilization cellulose **thiol** deriv;
cholinesterase immobilization cellulose **thiol** deriv; trypsin immobilization cellulose **thiol** deriv

IT Enzymes
RL: PROC (Process)
(immobilization of, on cellulose **thiol** derivs.)

IT 9000-81-1 9001-08-5 9002-07-7
RL: PROC (Process)
(immobilization of, on cellulose **thiol** derivs.)

IT 9001-08-5DP, mercaptobutylamidino deriv. 9002-07-7DP,
isothiocyanatopropyl deriv.
RL: SPN (Synthetic preparation); PREP (Preparation)
(prepn. and immobilization on cellulose **thiol** derivs.)

IT 9004-34-6DP, **thiol** derivs. 37324-27-9P
RL: SPN (Synthetic preparation); PREP (Preparation)
(prepn. and use in enzyme immobilization)

=> d L6 3 ibib

L6 ANSWER 3 OF 3 CAPLUS COPYRIGHT 2002 ACS
ACCESSION NUMBER: 1981:564692 CAPLUS
DOCUMENT NUMBER: 95:164692
TITLE: **Thiol** derivatives of cellulose as supports
for the immobilization of non-**thiol** enzymes
Gemeiner, Peter; Zemek, Jiri
AUTHOR(S):
CORPORATE SOURCE: Inst. Chem., Slovak Acad. Sci., Bratislava, 809 33,
Czech.
SOURCE: Collect. Czech. Chem. Commun. (1981), 46(7), 1693-700
CODEN: CCCCAK; ISSN: 0366-547X
DOCUMENT TYPE: Journal
LANGUAGE: English

=> s **thiol?** and pectin
67200 THIOL?
17317 PECTIN
4168 PECTINS
18878 PECTIN
(PECTIN OR PECTINS)

L7 18 THIOL? AND PECTIN

=> d L18 1-18 ti

L18 NOT FOUND

The L-number entered has not been defined in this session, or it
has been deleted. To see the L-numbers currently defined in this
session, enter DISPLAY HISTORY at an arrow prompt (=>).

=> d L7 1-18 ti

L7 ANSWER 1 OF 18 CAPLUS COPYRIGHT 2002 ACS
TI Adsorbent preparation, dry process for crosslinked polysaccharides, and
recovery and separation of arsenic ions using the same

L7 ANSWER 2 OF 18 CAPLUS COPYRIGHT 2002 ACS

TI Oral pharmaceutical preparation embedded in an oily matrix and coated by
enteric coating polymers

L7 ANSWER 3 OF 18 CAPLUS COPYRIGHT 2002 ACS
TI Kiwi protein inhibitor of **pectin** methylesterase. Amino-acid sequence and structural importance of two disulfide bridges

L7 ANSWER 4 OF 18 CAPLUS COPYRIGHT 2002 ACS
TI Purification and some properties of protease from *Actinidia eriantha* Benth

L7 ANSWER 5 OF 18 CAPLUS COPYRIGHT 2002 ACS
TI Keratinocyte growth factor-2 formulations for promotion of wound healing

L7 ANSWER 6 OF 18 CAPLUS COPYRIGHT 2002 ACS
TI Status of the **thiol**-dependent cytoprotectant systems under conditions of lead intoxication and high-**pectin** diet

L7 ANSWER 7 OF 18 CAPLUS COPYRIGHT 2002 ACS
TI Effect of mercurials on the activity of **pectin** methylesterase from egg plant (*Solanum melongena* L.)

L7 ANSWER 8 OF 18 CAPLUS COPYRIGHT 2002 ACS
TI Pharmaceutical compositions containing biologically active agents contained within a polymeric shell

L7 ANSWER 9 OF 18 CAPLUS COPYRIGHT 2002 ACS
TI Stabilized suspension of magnetic particles and its preparation and use in NMR diagnosis

L7 ANSWER 10 OF 18 CAPLUS COPYRIGHT 2002 ACS
TI Characterization of long-term extension of isolated cell walls from growing cucumber hypocotyls

L7 ANSWER 11 OF 18 CAPLUS COPYRIGHT 2002 ACS
TI Purification and properties of actinidin from *Actinidia chinensis*

L7 ANSWER 12 OF 18 CAPLUS COPYRIGHT 2002 ACS
TI An alkaline extracellular protease produced by *Cladosporium cucumerinum* and its possible importance in the development of scab disease of cucumber seedlings

L7 ANSWER 13 OF 18 CAPLUS COPYRIGHT 2002 ACS
TI Deodorization by aerobes

L7 ANSWER 14 OF 18 CAPLUS COPYRIGHT 2002 ACS
TI **Pectin** esterase(s) from sour oranges (*Citrus aurantium* Linn)

L7 ANSWER 15 OF 18 CAPLUS COPYRIGHT 2002 ACS
TI Rust remover containing a mercapto compound

L7 ANSWER 16 OF 18 CAPLUS COPYRIGHT 2002 ACS
TI Colorimetric determination of volatile sulfur compounds in foods. II. Reaction with bis(*p*-nitrophenyl) disulfide

L7 ANSWER 17 OF 18 CAPLUS COPYRIGHT 2002 ACS
TI Origin of methanol and dimethyl sulfide from cooked foods

L7 ANSWER 18 OF 18 CAPLUS COPYRIGHT 2002 ACS
TI Reductones derived from 3,4-dihydroxy-2,5-dicarboxylic esters of furan,

thiophene, N-phenylpyrrole, and selenophene

=> s thiol?(p)hyaluronic
67200 THIOL?
10110 HYALURONIC
1 HYALURONICS
10110 HYALURONIC
(HYALURONIC OR HYALURONICS)
L8 21 THIOL?(P)HYALURONIC

=> d L8 1-21 ti,kwic

L8 ANSWER 1 OF 21 CAPLUS COPYRIGHT 2002 ACS
TI Gold plating and biofunctionalization of ferromagnetic magnetic tweezers:
Application for local studies of soft surface-grafted polymer films
AB . . . resulting gold layer is a versatile platform for further
biofunctionalization using a wide variety of std. coupling protocols
based
on **thiol** chem. Several methods, such as electron microscopy,
elemental anal., x-ray powder diffraction, and XPS, have been used for
quant. and. . . of the coatings. The magnetic tweezers are used for
local quant. characterization of the elasticity of soft surface-grafted
films of **hyaluronic acid**. A method for the calibration of the
magnetization of each bead chosen for the measurement is introduced which
is. . .

L8 ANSWER 2 OF 21 CAPLUS COPYRIGHT 2002 ACS
TI Hyaluronic acid-protein conjugates, pharmaceutical compositions and
related methods
AB . . . present invention broadly relates to the field of protein
modification and, more specifically, the attachment of low mol. wt.,
derivatized **hyaluronic acid** polymer to proteins including leptin
and analogs thereof (the term "protein" as used herein is synonymous with
"polypeptide" or "peptide" unless otherwise indicated). The
hyaluronic acid-protein conjugates of the present invention
exhibit longer sustained blood levels than formulations contg. protein
alone, thus providing an important advantage in the therapeutic setting.
Conjugates of low mol. wt. sodium hyaluronate with free-**thiol**
osteroprotegrin was prep'd.

L8 ANSWER 3 OF 21 CAPLUS COPYRIGHT 2002 ACS
TI Synthesis of .beta.-D-glucopyranosyl(1.fwdarw.3)-1-**thiol**
-.beta.-glucosamine disaccharide derivative as building block for the
synthesis of **hyaluronic acid**
TI Synthesis of .beta.-D-glucopyranosyl(1.fwdarw.3)-1-**thiol**
-.beta.-glucosamine disaccharide derivative as building block for the
synthesis of **hyaluronic acid**
ST glucopyranosyl **thiolglucosamine** prepn **hyaluronic acid**
IT Glycosylation
(synthesis of .beta.-D-glucopyranosyl(1.fwdarw.3)-1-**thiol**
-.beta.-glucosamine, a building block for the synthesis of
hyaluronic acid)
IT 9004-61-9P, **Hyaluronic acid**
RL: PNU (Preparation, unclassified); PREP (Preparation)
(synthesis of .beta.-D-glucopyranosyl(1.fwdarw.3)-1-**thiol**
-.beta.-glucosamine, a building block for the synthesis of
hyaluronic acid)
IT 66-84-2, D-Glucosamine hydrochloride 76-03-9, Trichloroacetic acid,

reactions 83-87-4, D-Glucose pentaacetate 98-88-4, Benzoyl chloride 106-45-6, p-Thiocresol 116-11-0 123-76-2, Levulinic acid 1125-88-8, Benzaldehyde dimethyl acetal 17341-93-4

RL: RCT (Reactant)

(synthesis of .beta.-D-glucopyranosyl(1.fwdarw.3)-1-thiol
-.beta.-glucosamine, a building block for the synthesis of
hyaluronic acid)

IT 1152-39-2P 28244-94-2P 97562-23-7P 122210-01-9P 219518-19-1P
220645-20-5P 323195-38-6P 323195-39-7P 323195-40-0P 323195-41-1P
323195-42-2P 323195-43-3P 323195-44-4P 323195-45-5P 323195-46-6P
323195-47-7P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation)
(synthesis of .beta.-D-glucopyranosyl(1.fwdarw.3)-1-thiol
-.beta.-glucosamine, a building block for the synthesis of
hyaluronic acid)

IT 323195-37-5P

RL: SPN (Synthetic preparation); PREP (Preparation)
(synthesis of .beta.-D-glucopyranosyl(1.fwdarw.3)-1-thiol
-.beta.-glucosamine, a building block for the synthesis of
hyaluronic acid)

L8 ANSWER 4 OF 21 CAPLUS COPYRIGHT 2002 ACS

TI Oxidative damage of eye tissues and protection by thioctic acid

AB . . . in the mol. wt. compn. of bovine-lens homogenates mediated by illuminated riboflavin. Another indicator detected was the amt. of free thiol groups. Only the reduced form of thioctic acid (dihydrothioctic acid) protected from photo-oxidative changes, comparable to the synthetic dithiol dithiothreitol. . . by the oxidative fragmentation of .alpha.-keto-.gamma.-methiol-butyric acid (KMB) yielding ethylene. Both, thioctic acid and its reduced form, decreased ethylene formation. **Hyaluronic acid** is an important structural glycan in the vitreous. Oxidative degrdn. of **hyaluronic acid** by hypochlorous acid or Fenton-type oxidants is also diminished by thioctic acid, measured by mol.-wt. anal. of **hyaluronic acid**.

L8 ANSWER 5 OF 21 CAPLUS COPYRIGHT 2002 ACS

TI Thiol-containing biomaterials for medical and pharmaceutical use

IT 9004-34-6, Cellulose, biological studies 9004-54-0, Dextran, biological studies 9004-61-9, **Hyaluronic acid** 9005-25-8, Starch, biological studies 9007-28-7, Chondroitin sulfate 9012-76-4, Chitosan 26780-50-7, Poly(glycolide-lactide)

RL: DEV (Device component use); THU (Therapeutic use); BIOL (Biological study); USES (Uses)

(thiol-contg. biomaterials for medical and pharmaceutical use)

L8 ANSWER 6 OF 21 CAPLUS COPYRIGHT 2002 ACS

TI Synthetic polysulfated hyaluronic acid is a potent inhibitor for tumor necrosis factor production

AB . . . blocks for synthesizing nontoxic drugs for suppression of tumor necrosis factor (TNF) prodn. by inflammatory cells, we have chem. modified

hyaluronic acid (HA) and tested its effects in blocking TNF-.alpha. and TNF-.beta. prodn. in vitro. HA was chosen mainly for its.

. . . decreasing the extent of polysulfation, the inhibitory effect of HAS on TNF-.alpha. prodn. was diminished. Other chem. modifications, including deacetylation, **thiolation**, or redn. of the carboxylic groups, could not increase the efficacy of HA in suppression of

TNF-.alpha. prodn. Naturally polysulfated. . .

L8 ANSWER 7 OF 21 CAPLUS COPYRIGHT 2002 ACS
TI Inhibitory actions of **thiol** compounds in HOCl induced degradation of **hyaluronic acid**
TI Inhibitory actions of **thiol** compounds in HOCl induced degradation of **hyaluronic acid**
AB Effects of **thiol** compds. N-(N-L-.gamma.-glutamyl-L-cysteinyl)glycine, N-(2-mercaptopropionyl)glycine, and cysteine on the degrdn. of **hyaluronic acid** were investigated. Scavenging actions of **thiol** compds. on HOCl were examd.

L8 ANSWER 8 OF 21 CAPLUS COPYRIGHT 2002 ACS
TI Endothelial cell stimulation of smooth muscle glycosaminoglycan synthesis can be accounted for by transforming growth factor beta activity
AB . . . Here it is shown that the factor responsible is transforming growth factor beta (TGF-.beta.), as assessed by (1) proteinase and **thiol** sensitivity, (2) heat and acid enhancement of ECCM activity, and (3) neutralization of ECCM activity by anti-TGF-.beta. Ig. Anti-TGF-.beta. neutralization. . . showed that ECCM from EC of varying densities stimulated individual GAG to varying degrees. ECCM from low-d. EC preferentially stimulated **hyaluronic acid** (HA), whereas ECCM from intermediate- and high-d. cultures stimulated increasing amts. of sulfated GAG. Exposure of SMC to varying . . .

L8 ANSWER 9 OF 21 CAPLUS COPYRIGHT 2002 ACS
TI The mechanism of chondrocyte hydrogen peroxide damage. Depletion in intracellular ATP due to suppression of glycolysis caused by oxidation of glyceraldehyde-3-phosphate dehydrogenase
AB . . . be due to the oxidative inactivation of glyceraldehyde-3-phosphate dehydrogenase (G-3-PDH). Apparently, intrachondrocyte oxidant damage occurs through oxidn. of the sensitive **thiol** (-SH) residue at the active center of G-3-PDH, with subsequent redn. in the rate of glycolytic ATP synthesis and the intracellular concn. of ATP which is required for DNA, protein, proteoglycan, and **hyaluronic acid** synthesis.

L8 ANSWER 10 OF 21 CAPLUS COPYRIGHT 2002 ACS
TI **Hyaluronic** acid degrading reactions under hyperthermic conditions and effects of **thiols**
TI **Hyaluronic** acid degrading reactions under hyperthermic conditions and effects of **thiols**
AB **Thiols** promoted the formation of highly oxidizing **hyaluronic** acid (HA)-degrading species in the presence of chelated (EDTA) Fe3+, Fe2+, and Cu2+ compds. Hyperthermia (40-44.degree.) led to an addnl. increase in the prodn. of HA-degrading species by the interaction of **thiols** with EDTA-complexed transition metals. The most efficient **thiols** were cysteamine and cysteine; GSH, dithiothreitol, and N-acetylcysteine promoted the generation of the highly reactive radicals to a minor extent.
IT Transition metals, compounds
RL: BIOL (Biological study)
 (EDTA complexes, oxygen radical formation in presence of, hyperthermia and **thiols** effects on, **hyaluronic** acid degrdn. in relation to)
IT Fever and Hyperthermia
 Thiols, biological studies

RL: BIOL (Biological study)
(oxygen radical formation in presence of transition metal-EDTA complexes response to, **hyaluronic acid** degrdn. in relation to)

IT 7782-44-7D, radicals
RL: FORM (Formation, nonpreparative)
(formation of, **thiols** promotion of, hyperthermia effect on, **hyaluronic acid** degrdn. in relation to)

IT 60-00-4D, transition metal complexes 7439-89-6D, EDTA complexes
7440-50-8D, EDTA complexes 7447-39-4D, EDTA complexes 7705-08-0D,
EDTA complexes 7720-78-7D, EDTA complexes

RL: BIOL (Biological study)
(oxygen radical formation in presence of, hyperthermia and **thiols** effects on, **hyaluronic acid** degrdn. in relation to)

L8 ANSWER 11 OF 21 CAPLUS COPYRIGHT 2002 ACS

TI Copper dependent control of the enzymic and phagocyte induced degradation of **hyaluronic acid**, synovial fluid and cytochrome c

AB Studies on the oxidn. of Cu-thionein by xanthine oxidase are described in detail. Cu²⁺ was released by Cu(I)-**thiolate** (chromophore of Cu-thionein) degrdn. Complete but reversible inhibition of xanthine oxidase occurred in the presence of 5 .mu.M Cu; diminished xanthine oxidase inhibition occurred with the Cu(I)-**thiolate** chromophore of yeast Cu-thionein. The xanthine oxidase-dependent oxidn. of xanthine or hypoxanthine degraded **hyaluronic acid** (a component of synovial fluid), and Cu-thionein and CuSO₄ inhibited the degrdn. Effects on bovine synovial fluid also are. . . c by phagocyte-generated hypochlorite was a measure of polymorphonuclear leukocyte (PMN) activity. Activated PMNs produce excited O species which depolymerize **hyaluronic acid** but no degrdn. was obsd. on the presence of ceruloplasmin and CuSO₄; some degrdn. was obsd. with Cu-thionein. Relations. . .

L8 ANSWER 12 OF 21 CAPLUS COPYRIGHT 2002 ACS

TI Radiation induced depolymerization of **hyaluronic acid** (HA) in aqueous solutions at pH 7.4

AB Radiolytic depolymn. of **hyaluronic acid** (HA, a heteropolysaccharide) in aq. solns. under a variety of conditions demonstrates that the damaging effect of radiolytic radical. . . order OH.cntdot.>e-aq>O-2. Cysteine, penicillamine, and dithiothreitol protected against primary radiolytic species. The enzyme superoxide dismutase (SOD) and the above 3 **thiols** do not protect against the radiolytic species generated by the .lambda.-irradn. of aerated Na formate solns. The results also indicate that the reaction between CO-2 anion and **hyaluronic acid** is faster than the reaction between O-2 and **hyaluronic acid** and that CO-2 anions are not scavenged by superoxide dismutase. The results further suggest that tert-BuOH radicals interact with **hyaluronic acid** and reduce the viscosity of HA solns. Preliminary pulse radiolysis expts. do demonstrate a reaction between CO-2 radical and **hyaluronic acid**.

L8 ANSWER 13 OF 21 CAPLUS COPYRIGHT 2002 ACS

TI Connective tissue activation. XXVIII. A connective tissue activating peptide from human urine

AB . . . be different from EGF and IgG. This urinary connective tissue-activating factor (CTAP-U) [92307-84-1] is nondialyzable, labile to protease, stable to **thiols**, heat, and acid, and has an acidic isoelec. point. Purified prepns. of CTAP-U have biol. activities that

cause human connective tissue cells to synthesize incremental amts. of ¹⁴C-labeled **hyaluronic acid** [9004-61-9], [35S]proteoglycans, and [3H]DNA in vitro. The cell spectrum responsive to this substance includes human synovial cells, human chondrocytes, . . .

L8 ANSWER 14 OF 21 CAPLUS COPYRIGHT 2002 ACS
TI Cartilage-degrading neutral proteinase secreted by Yoshida sarcoma cells. Purification and properties
AB . . . help of a new assay system for measurement of proteoglycan core protein degrdn., which utilizes aminopropyl glass beads derivatized with **hyaluronic acid**. This enzyme, with a neutral pH optimum and apparent mol. wt. of .apprx.30,000, was secreted into culture medium in. . . form. It was resistant to cartilage-derived inhibitors and to .alpha.2-macroglobulin as well as to synthetic and natural inhibitors of serine, **thiol**, and carboxyl proteinases. It was inhibited by 1,10-phenanthroline and **thiols** at relatively high concns., and therefore is probably a metalloproteinase. The enzyme degraded type V collagen, types I and II. . .

L8 ANSWER 15 OF 21 CAPLUS COPYRIGHT 2002 ACS
TI Primary selection and study of the properties of chemical agents for protection from radiation injury
AB . . . presented for a primary selection and study of radioprotectors, and applied in studying the protective effects on erythrocytes of aminoalkyl **thiols**, aminoalkyl disulfides, aminoalkylisothiuronium compds., Bunte salts, and thiazolidines. A radiomimetic effect of lipoidal toxic compds. produced during irradn. is significantly reduced when phospholipids from animal tissues are added. Polysaccharides except **hyaluronic acid** exhibited a similar action in the erythrocyte radiomimetic model to that of phospholipids.

L8 ANSWER 16 OF 21 CAPLUS COPYRIGHT 2002 ACS
TI Inhibition of steroid .DELTA.4-reductase by polysulfated polysaccharides and para-chloromercuribenzoic acid
AB Four heparinoids and 2 naturally occurring, high mol. wt. polysaccharides with varying contents, including heparin, chondroitin sulfate, and **hyaluronic acid**, were assayed for inhibitory effects on rat liver steroid .DELTA.4-reductase activity. All active compds. had a 15-18% S content, . . . of cortisone reductase of only 60%, while p-chloromercuribenzoate (PCMB) (5 .times. 10⁻⁴M) completely inhibited the enzyme, indicating the importance of **thiol** groups to enzymic activity. PCMB and heparin were additive in a manner which gave greater inhibition than could be demonstrated. . .

L8 ANSWER 17 OF 21 CAPLUS COPYRIGHT 2002 ACS
TI Depolymerization of hyaluronic acid by autoxidants and radiations
AB The effects of x-rays, .beta.-rays (32P internally), and autoxidants were compared for the aerobic depolymerization of **hyaluronic acid**. A 0.02mM soln. of ascorbic acid reduced the viscosity of **hyaluronic acid** solns. at about the same rate as 2400 rads of x-rays or .beta.-rays. The yields (G values) were low, about 0.015. Visible light also caused the degradation of **hyaluronic acid** in the presence of riboflavin as a sensitizer. A survey of compds. with possible autoxidant properties showed that the following structures were assocd. with such activity: enediols (reduced or oxidized), quinones and hydroquinones, many **thiols** (not disulfides), m-nitrophenol, and a few metallic ions, esp. Cu⁺, Fe⁺⁺, and Sn⁺⁺. Antioxidant activity of these compds. was tested extensively for the system **hyaluronic acid**-ascorbic acid,

and some compds. were studied in the radiation systems. The effects in both systems were similar, but were. . .

L8 ANSWER 18 OF 21 CAPLUS COPYRIGHT 2002 ACS
TI Inactivation of solutions of transforming deoxyribonucleic acid by gamma.-rays in the presence of protective substances
AB . . . had a protective effect. The rate of I inactivation was unaffected by irradiation rates of 2.4, 8.3, and 33 kr./hr.; **hyaluronic acid** from human umbilical cord, RNA, or yeast ext. used at 1 mg./cc. had a protective effect. The protective effect. . . the original, were 1, 8, 10, 17, 57, 170, 76, 330, 719, 92, and 560 kr./hr. for control; 1 mg./cc. **hyaluronic acid**, yeast RNA, or yeast ext. 0.25 and 1.0 mg./cc. thiourea; 0.25, 1.0, and 2.5 mg./cc. cysteine; and 0.25 and 1.0 mg./cc. mercaptoethanol, resp. The protective effect of **thiols** is considered specific and different from that of the yeast ext.

L8 ANSWER 19 OF 21 CAPLUS COPYRIGHT 2002 ACS
TI Depolymerization of hyaluronic acid by the oxidative-reductive depolymerization (ORD) reaction
AB cf. CA 54, 4692d, 24922g. **Hyaluronic acid** (I) was depolymerized in vitro by various biol. occurring reducing agents, such as ferrous ions, ascorbic acid, hydroquinones, and **thiol** compds. Depolymerization was followed by the fall in intrinsic viscosity of 40 mg. % solns. of I in 0.2M phosphate. . .

L8 ANSWER 20 OF 21 CAPLUS COPYRIGHT 2002 ACS
TI State of polysaccharides (hyaluronic acid) in rheumatism
AB . . . the specific enzyme hyaluronidase is increased. An accumulation of nonspecific substances takes place in rheumatic patients. They cause destruction of **hyaluronic acid** and non-specific substances are excreted with the urine. These substances are found also in the urine of normal persons but in a much smaller amt. These nonspecific substances include ascorbic acid, diazo compds. and azoproteins, lecithin, **thiolactic acid**, some P compds., etc. Following treatment with salicylates the activity of hyaluronidase is depressed and the amt. of nonspecific. . .

L8 ANSWER 21 OF 21 CAPLUS COPYRIGHT 2002 ACS
TI Diffusing factors. The hyaluronidase activity of testicular extracts, bacterial culture filtrates and other agents that increase tissue permeability
AB The assocn. between hyaluronidase activity (ability to hydrolyze **hyaluronic acid**, a mucopolysaccharide present in certain mucoproteins) and diffusing factors from various sources was described. The reduction of viscosity and. . . oxidation products also possessed both these properties. Reducing substances were not formed in either case. Other reducing substances such as **thiolacetic acid**, H₂S, hydroquinone, pyrogallol, Na₂SO₃ and metol also decreased the viscosity of mucoprotein and had diffusing activity when injected. The. . .

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L8 ANSWER 5 OF 21 CAPLUS COPYRIGHT 2002 ACS
ACCESSION NUMBER: 1995:774579 CAPLUS
DOCUMENT NUMBER: 123:208920

TITLE: Thiol-containing biomaterials for medical and pharmaceutical use
 INVENTOR(S): Constancis, Alain; Soula, Gerard
 PATENT ASSIGNEE(S): Flamel Technologies, Fr.
 SOURCE: Fr. Demande, 28 pp.
 CODEN: FRXXBL
 DOCUMENT TYPE: Patent
 LANGUAGE: French
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
FR 2707992	A1	19950127	FR 1993-9198	19930721
FR 2707992	B1	19951013		
WO 9503272	A1	19950202	WO 1994-FR914	19940721
W: JP, US				
RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE				
EP 710226	A1	19960508	EP 1994-922288	19940721
EP 710226	B1	19981014		
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LI, LU, MC, NL, PT, SE				
JP 09503490	T2	19970408	JP 1994-504980	19940721
AT 172191	E	19981015	AT 1994-922288	19940721
US 5646239	A	19970708	US 1996-578539	19960306
PRIORITY APPLN. INFO.:			FR 1993-9198	19930721
			WO 1994-FR914	19940721
OTHER SOURCE(S):	MARPAT 123:208920			

L8 ANSWER 6 OF 21 CAPLUS COPYRIGHT 2002 ACS
 ACCESSION NUMBER: 1994:570103 CAPLUS
 DOCUMENT NUMBER: 121:170103
 TITLE: Synthetic polysulfated hyaluronic acid is a potent inhibitor for tumor necrosis factor production
 AUTHOR(S): Chang, Nan Shan; Intrieri, Catherine; Mattison, Jeffery; Armand, Gerard
 CORPORATE SOURCE: Lab. Mol. Immunol., Guthrie Res. Inst., Sayre, PA,
 USA
 SOURCE: J. Leukocyte Biol. (1994), 55(6), 778-84
 DOCUMENT TYPE: CODEN: JLBIET; ISSN: 0741-5400
 LANGUAGE: English

COST IN U.S. DOLLARS	SINCE FILE ENTRY	TOTAL SESSION
FULL ESTIMATED COST	92.21	92.36
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE	TOTAL